Chemical potential shift induced by double-exchange and polaronic effects in Nd_{1-r}Sr_rMnO₃

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We have studied the chemical potential shift as a function of temperature in $Nd_{1-x}Sr_xMnO_3$ by measurements of core-level photoemission spectra. For ferromagnetic samples (x=0.4 and 0.45), we observed an unusually large upward chemical potential shift with decreasing temperature in the low-temperature region of the ferromagnetic metallic phase. This can be explained by the double-exchange (DE) mechanism if the e_g band is split by the dynamical and/or local Jahn-Teller effect. The shift was suppressed near the Curie temperature (T_C), which we attribute to the crossover from the DE to lattice-polaron regimes.

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I. INTRODUCTION

Perovskite-type manganites of the chemical formula $R_{1-x}A_x$ MnO₃ (where R is a rare earth and A is an alkalineearth metal) have attracted much interest because of their colossal magnetoresistance.^{1,2} Historically, the magnetic and transport properties of this kind of materials have been understood in terms of double-exchange (DE) interaction between the localized t_{2g} electrons and the itinerant e_g electrons.3-5 In this model, the kinetic energy gain of the holes doped into the e_g band is maximized when the spins of the t_{2g} electrons are aligned in the same direction, thereby stabilizing the ferromagnetic ground state. However, it has been pointed out by Millis et al.⁶ that the DE model is insufficient to explain the huge change in the resistivity under a magnetic field and the high resistivity above the Curie temperature (T_c) . They have proposed a model including the dynamical Jahn-Teller (JT) distortion of the MnO₆ octahedra in addition to the DE interaction and reproduced the experimentally observed behavior of the resistivity. It has also been proposed that the extraordinary enhancement of the resistivity in the manganites may result from the emergence of lattice polarons in the paramagnetic insulating (PI) phase.⁶⁻⁹ Koo et al.¹⁰ have found that lattice polarons in Nd_{0.7}Sr_{0.3}MnO₃ are strongly suppressed by applying magnetic field and do not completely disappear at high fields, corresponding to an admixture of the "conducting" and "insulating" carriers. From x-ray scattering and neutron scattering studies, ^{7–9} indeed, the diffuse scattering due to the lattice polarons has been observed in the high-temperature PI phase of manganites and gradually disappears with decreasing temperature in the ferromagnetic metallic (FM) phase. On the other hand, several local structural studies have suggested that a dynamical or local JT distortion persists in the FM phase too. Extended x-ray absorption fine structure (EXAFS) studies have shown that the lattice distortion of the MnO₆ octahedra is found in the FM phase of La_{1-x}Ca_xMnO₃ at low temperatures.^{11,12} From the pair-density function analysis of pulsed neutron diffraction data, the local JT distortion has been observed in the FM phase of $La_{1-x}Sr_xMnO_3$.¹³ If the JT distortion exists in the FM phase, the degeneracy of the e_g band may be lifted already in that phase.

In order to obtain insight into the competition between the DE mechanism and the lattice-polaron effect, the measurement of chemical potential shift as a function of temperature gives much insight. According to the DE model, a temperature-dependent chemical potential shift occurs due to the change in the e_g bandwidth, as schematically shown in Fig. $1.^{14}$ If the e_g band is split by JT distortion and the oneorbital DE model becomes relevant, the chemical potential is shifted upward for hole concentration x < 0.5 with decreasing temperature and downward for x > 0.5. ¹⁴ If one takes into account the double degeneracy of the e_{σ} orbitals [Fig. 1(a)], a downward shift with decreasing temperature would be expected in the FM phase for $0 \le x \le 1$ because the up-spin band of the e_g orbitals is less than half-filled in the $R_{1-x}A_x$ MnO₃ compounds. Therefore, the temperaturedependent chemical potential shift is sensitive to the splitting of the e_p band and therefore to the dynamical and/or local JT effect. Schulte et al. 15 have investigated the change of the work function in La_{1.2}Sr_{1.8}Mn₂O₇ as a function of temperature by measurements of photoemission spectra and attributed the change to the temperature-dependent shift of the chemical potential. Alternatively, the chemical potential shift can be deduced from the shifts of photoemission spectra because the binding energies of the spectra are measured relative to the chemical potential. We employ the latter method in this work.

 $Nd_{1-x}Sr_xMnO_3$ (NSMO) is a suitable system for clarifying the relationship between the DE interactions and the existence of lattice polarons because it shows the FM phase for $x \le 0.5$ and the so-called charge-exchange-type antiferromagnetic charge-ordered phase in the doping region close to the half-doping x=0.5, as shown in Fig. 2. ^{1,16} In this work, we study the chemical potential shift in NSMO as a function of

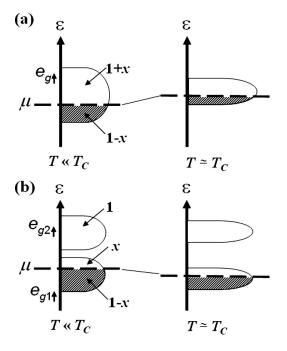


FIG. 1. Schematic pictures of the temperature-dependent density of states due to the DE interaction in FM phase for x < 0.5. (a) Degenerate two-orbital model; (b) One-orbital model resulting from the Jahn-Teller splitting. μ , T_C , and x denote the chemical potential, the Curie temperature, and the hole concentration, respectively.

temperature by measurements of core-level photoemission spectra. We found that the chemical potential shift was large in the low-temperature part of the FM phase as predicted by the DE model and dynamical and/or local JT effect and was suppressed at high temperatures near T_{C} . We consider that

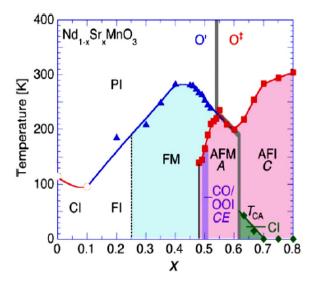


FIG. 2. (Color online) Electronic phase diagram of $Nd_{1-x}Sr_xMnO_3$. PI: paramagnetic insulating phase; CI: spin-canted insulating phase; FM: ferromagnetic metallic phase; FI: ferromagnetic insulating phase; CO/OOI: charge-orbital ordered insulating phase; AFM: antiferromagnetic metallic phase; AFI: antiferromagnetic insulating phase (Refs. 1 and 16).

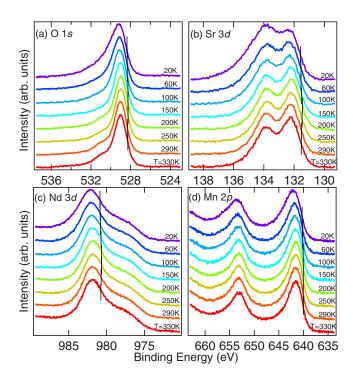


FIG. 3. (Color online) Core-level photoemission spectra of $Nd_{0.6}Sr_{0.4}MnO_3$ taken with the Mg $K\alpha$ line. (a) O 1s, (b) Sr 3d, (c) Nd 3d, and (d) Mn 2p.

the different behaviors with temperature are related to the competition between the DE interaction and the lattice-polaron effect.

II. EXPERIMENT

Single crystals of NSMO (x=0.4 and 0.45) were prepared by the floating zone method. ¹⁶ X-ray photoemission spectroscopy measurements were performed using the photon energies of $h\nu$ =1253.6 eV (Mg $K\alpha$). All the photoemission measurements were performed under the base pressure of \sim 10⁻¹⁰ Torr at 20–330 K. The samples were repeatedly scraped *in situ* with a diamond file to obtain clean surfaces. The cleanliness of the sample surface was checked by the reduction of the shoulder on the high binding-energy side of the O 1s core level. Photoelectrons were collected using a Scienta SES-100 electron-energy analyzer. The energy resolution was about 800 meV. The measured binding energies were stable because the gold $4f_{7/2}$ core-level spectrum did not change in the measurements with the accuracy of \pm 10 meV at each temperature.

III. RESULTS AND DISCUSSION

In Fig. 3, we have plotted the spectra of the O 1s, Sr 3d, Nd 3d, and Mn 2p core levels in NSMO with x=0.4. The vertical lines mark the estimated positions of the core levels used in the present study. We employed the midpoint of the low binding-energy slope for the O 1s core level because the line shape on the higher-binding-energy side of the O 1s spectra is known to be affected by surface contamination or degradation. We also employed the midpoint for the Sr 3d

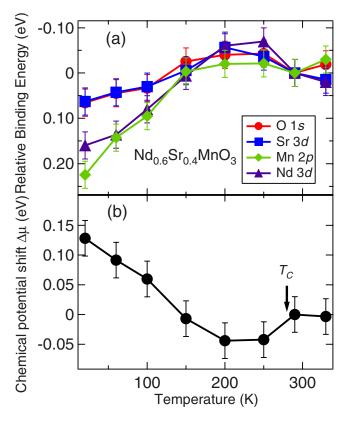


FIG. 4. (Color online) Core-level shifts and chemical potential shift in Nd_{0.6}Sr_{0.4}MnO₃. (a) Binding-energy shifts of the O 1s, Sr 3d, Nd 3d, and Mn 2p core levels as functions of temperature relative to 290 K. (b) Chemical potential shift $\Delta\mu$ as a function of temperature. T_C denotes the Curie temperature.

and Mn 2p core levels. As for the Nd 3d core level, 80% of the peak height of the low-binding-energy slope was used because the line shape near the midpoint on the lower-binding-energy side slightly changed with temperature.

Figure 4(a) shows the binding-energy shift of each core level as a function of temperature for NSMO with x=0.4. One can see that the observed binding-energy shifts with temperature were approximately common to the O 1s, Sr 3d, Nd 3d, and Mn 2p core levels. In order to deduce the chemical potential shift from the set of core-level data, we utilize the formula^{17,18} that the shift ΔE_B of the binding energy is given by $\Delta E_B = \Delta \mu + K \Delta Q + \Delta V_M - \Delta E_R$, where $\Delta \mu$ is the change in the chemical potential, K is the Coulomb coupling constant between the valence and core electrons, ΔQ is the change in the number of valence electrons on the atom considered, ΔV_M is the change in the Madelung potential, and ΔE_R is the change in the extra-atomic relaxation energy. The ΔE_R term is known to be neglected from the main origin of the core-level shifts in transition-metal oxides. 19,20 Also, the change in the Madelung potential has negligible effects on the core-level shifts because the identical core-level shifts were observed. In the measurements of the dopingdependent chemical potential shift in Pr_{1-x}Ca_xMnO₃, all the core levels have shown identical shifts with hole concentration except for the Mn 2p core level, where the effect of chemical shift is superimposed.²¹ Here, all the core levels including the Mn 2p core level exhibit similar shifts with temperature in contrast to the measurements of doping-dependent core-level shifts. Therefore, we assume that the shifts of the core levels are largely due to the chemical potential shift and take the average of the shifts of the four core levels as a measure of $\Delta\mu$ as a function of temperature in NSMO.

Figure 4(b) shows the temperature-dependent chemical potential shift in NSMO with x=0.4. We observed a large upward chemical potential shift with decreasing temperature in the FM phase of NSMO at low temperatures. Furukawa¹⁴ proposed an anomalous temperature-dependent chemical potential shift below T_C due to the DE interaction through the change of the e_g bandwidth with temperature. He also predicted that the magnitude of the shift was estimated to be about 0.1 eV when the e_g bandwidth was of the order of ~ 1 eV. ¹⁴ If the e_g band remains degenerate in the FM phase of NSMO, one would expect to see a downward chemical potential shift with decreasing temperature based on the DE interaction because the up-spin band of the e_g orbitals is less than half-filled for $0 \le x \le 1$. On the other hand, if the degeneracy of the e_g band is lifted by the dynamical and/or local JT distortion, the upward chemical potential shift with decreasing temperature is predicted because the band is more than half-filled for x < 0.5 (see Fig. 1). Therefore, we attribute the large upward shift of the chemical potential with decreasing temperature in the low-temperature FM phase of NSMO to the change in the width of the JT-split e_{ρ} band caused by the DE mechanism. The magnitude of the observed shift is in quantitative agreement with the results of the one-orbital DE model. 14

We consider that the temperature-dependent splitting of the e_g band may also influence the temperature-dependent chemical potential shift in the FM phase of NSMO. The intensity of the diffuse scattering due to the JT effect gradually increases with increasing temperature in the FM phase near T_C , which may indicate further increase of the energy splitting of the e_g level.⁷⁻⁹ EXAFS studies have indeed indicated that the transition from the PI phase to the FM phase involves a decrease of the JT distortion but that the magnitude of the JT distortion remains substantial and becomes nearly temperature independent within the FM phase.11 If temperature dependence of the JT distortion was the dominant cause of the temperature-dependent chemical potential shift, the temperature-dependent shift would be even stronger at higher temperatures for the one-orbital case and the suppression of chemical potential shift in NSMO within the FM phase near T_C cannot be explained. At high temperatures near T_C in the PI phase, the diffuse scattering due to the formation of lattice polarons has been observed by means of x-ray scattering and neutron scattering studies.^{7–9} We consider that the suppression of the shift in the FM phase at high temperatures is connected with the influence of the lattice polarons and the DE model is no more effective at those high temperatures.

In Fig. 5, we compare $\Delta\mu$ for NSMO (x=0.4) with the shifts of the O 1s core level (in the process of increasing temperature) and the valence band of La_{1-x}Sr_xMnO₃ (x=0.2, 0.3, and 0.4), $\Delta\mu$ for bilayered system La_{2-2x}Sr_{1+2x}Mn₂O₇ (x=0.4) and correlated hopping model of Mn³⁺/Mn⁴⁺ mixed-valence state as a function of

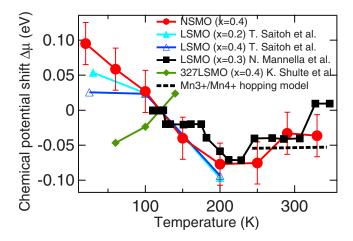


FIG. 5. (Color online) Comparison of the chemical potential shift $\Delta\mu$ in $\mathrm{Nd}_{1-x}\mathrm{Sr}_x\mathrm{MnO}_3$ (x=0.4) with the shifts of the O 1s core level (in the process of increasing temperature) and the valence band of $\mathrm{La}_{1-x}\mathrm{Sr}_x\mathrm{MnO}_3$ (x=0.2, 0.3, and 0.4 with $T_C\approx$ 310, 370, and 370 K, respectively) (Refs. 22 and 23), $\Delta\mu$ for the bilayered system $\mathrm{La}_{2-2x}\mathrm{Sr}_{1+2x}\mathrm{Mn}_2\mathrm{O}_7$ (x=0.4 with $T_C\approx$ 125 K) (Ref. 15) and correlated hopping model of $\mathrm{Mn}^{3+}/\mathrm{Mn}^{4+}$ mixed-valence state. The spectral shifts have been translated to the chemical potential shift, assuming that the spectral shifts are primarily close to the chemical potential shift.

temperature. ^{15,22–26} We consider that the temperature-dependent shifts of the O 1s core level and the valence band for La_{1-x}Sr_xMnO₃ are ascribed to the temperature-dependent change in $\Delta\mu$. For NSMO and La_{1-x}Sr_xMnO₃, the $\Delta\mu$ curves show similar temperature dependences in the sense that the chemical potential is shifted upward with decreasing temperature. We consider that the shifts are understood in terms of the DE interaction (and possibly the temperature dependence of the splitting of the e_g band caused by the dynamical and/or local JT effect). ^{7–9,14} However, the $\Delta\mu$ for the bilayered compound La_{2-2x}Sr_{1+2x}Mn₂O₇ reported by Schulte *et al.* ¹⁵ is opposite to the prediction of the DE model. They speculated that the two dimensionality might affect the temperature-dependent chemical potential shift, but its origin remains an open question.

Also, we have compared the experimentally observed shifts with the results of the correlated hopping model in the splitting of the e_g band due to the dynamical and/or local JT distortion. $^{24-26}$ The correlated hopping model is given by $\Delta \mu = -k_B \ln(\frac{g_3}{g_4}\frac{x}{1-x})T + \text{const}$, where g_3 and g_4 are the spinorbital degeneracies of Mn³⁺ and Mn⁴⁺, respectively, and x is the fraction of Mn⁴⁺ ions. $^{24-26}$ The parameters were fixed at $g_3 = 5$, $g_4 = 4$, and x = 0.4, corresponding to the Mn³⁺/Mn⁴⁺ mixed-valence state in the split of the e_g band. We have plotted the calculated $\Delta \mu$, as shown by a dashed line in Fig. 5. $^{24-26}$ The result is in qualitative agreement with the observed chemical potential shift in the high-temperature region of NSMO with x = 0.4.

For other hole concentrations of NSMO, too, we confirmed that the observed temperature-dependent shifts were

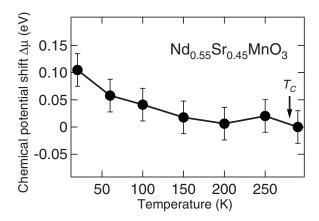


FIG. 6. Chemical potential shift in $Nd_{0.55}Sr_{0.45}MnO_3$. T_C denotes the Curie temperature.

common to the O 1s, Sr 3d, Nd 3d, and Mn 2p core levels and deduced the temperature-dependent chemical potential shifts by taking the average of the shifts of the four core levels. Figure 6 shows the $\Delta\mu$ for NSMO with x=0.45 as a function of temperature. The shift $\Delta\mu$ was suppressed just below the T_C as in the case of x=0.4, in accordance with the enhancement of the diffuse scattering due to the lattice polarons. $^{7-9}$ We attribute the shift in the low-temperature region of the FM phase for x=0.45 to the DE interaction (and the temperature dependence of the splitting of the e_g band due to the dynamical and/or local JT effect) as in the case of x=0.4. $^{7-9,14}$ The magnitude of the observed shift for x=0.45 is smaller than that for x=0.4, consistent with the one-orbital DE model resulting from the JT splitting [see Fig. 1(b)].

IV. CONCLUSION

We have measured the chemical potential shift as a function of temperature in NSMO by means of core-level photoemission spectroscopy. We have found an anomalous upward chemical potential shift with decreasing temperature in the low-temperature region of the FM phase and its suppression in the high-temperature region of the FM phase near T_C . We attribute the large shift in the low-temperature region to the change of bandwidth due to the DE interaction (and possibly the temperature dependence of the splitting of the e_g level caused by the dynamical and/or local JT effect). Also, the suppression of the shift at higher temperatures is ascribed to the influence of lattice-polaron formation.

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- ¹Y. Tokura, Rep. Prog. Phys. **69**, 797 (2006).
- ²E. Dagotto, Nanoscale Phase Separation and Colossal Magnetoresistance, Springer Series in Solid State Sciences Vol. 136 (Springer, Berlin, 2002).
- ³C. Zener, Phys. Rev. **82**, 403 (1951).
- ⁴P. W. Anderson and H. Hasegawa, Phys. Rev. **100**, 675 (1955).
- ⁵N. Furukawa, J. Phys. Soc. Jpn. **63**, 3214 (1994).
- ⁶ A. J. Millis, B. I. Shraiman, and R. Mueller, Phys. Rev. Lett. 77, 175 (1996).
- ⁷S. Shimomura, N. Wakabayashi, H. Kuwahara, and Y. Tokura, Phys. Rev. Lett. **83**, 4389 (1999).
- ⁸ V. Kiryukhin, T. Y. Koo, A. Borissov, Y. J. Kim, C. S. Nelson, J. P. Hill, D. Gibbs, and S.-W. Cheong, Phys. Rev. B 65, 094421 (2002).
- ⁹ P. Dai, J. A. Fernandez-Baca, N. Wakabayashi, E. W. Plummer, Y. Tomioka, and Y. Tokura, Phys. Rev. Lett. 85, 2553 (2000).
- ¹⁰T. Y. Koo, V. Kiryukhin, P. A. Sharma, J. P. Hill, and S.-W. Cheong, Phys. Rev. B **64**, 220405(R) (2001).
- ¹¹ A. Lanzara, N. L. Saini, M. Brunelli, F. Natali, A. Bianconi, P. G. Radaelli, and S.-W. Cheong, Phys. Rev. Lett. 81, 878 (1998).
- ¹²C. H. Booth, F. Bridges, G. H. Kwei, J. M. Lawrence, A. L. Cornelius, and J. J. Neumeier, Phys. Rev. Lett. 80, 853 (1998).
- ¹³D. Louca, T. Egami, E. L. Brosha, H. Röder, and A. R. Bishop, Phys. Rev. B **56**, R8475 (1997).
- ¹⁴N. Furukawa, J. Phys. Soc. Jpn. **66**, 2523 (1997).
- ¹⁵ K. Schulte, M. A. James, L. H. Tjeng, P. G. Steeneken, G. A. Sawatzky, R. Suryanarayanan, G. Dhalenne, and A. Revcolevs-

- chi, Phys. Rev. B 64, 134428 (2001).
- ¹⁶H. Kuwahara, Y. Tomioka, Y. Moritomo, and Y. Tokura, Science 270, 961 (1995).
- ¹⁷S. Hüfner, *Photoelectron Spectroscopy* (Springer-Verlag, Berlin, 2003)
- ¹⁸ A. Fujimori, A. Ino, J. Matsuno, T. Yoshida, K. Tanaka, and T. Mizokawa, J. Electron Spectrosc. Relat. Phenom. **124**, 127 (2002).
- ¹⁹ A. Ino, T. Mizokawa, A. Fujimori, K. Tamasaku, H. Eisaki, S. Uchida, T. Kimura, T. Sasagawa, and K. Kishio, Phys. Rev. Lett. 79, 2101 (1997).
- ²⁰N. Harima, J. Matsuno, A. Fujimori, Y. Onose, Y. Taguchi, and Y. Tokura, Phys. Rev. B **64**, 220507(R) (2001).
- ²¹ K. Ebata, H. Wadati, M. Takizawa, A. Fujimori, A. Chikamatsu, H. Kumigashira, M. Oshima, Y. Tomioka, and Y. Tokura, Phys. Rev. B **74**, 064419 (2006).
- ²²T. Saitoh, A. Sekiyama, K. Kobayashi, T. Mizokawa, A. Fujimori, D. D. Sarma, Y. Takeda, and M. Takano, Phys. Rev. B **56**, 8836 (1997).
- ²³N. Mannella, A. Rosenhahn, C. H. Booth, S. Marchesini, B. S. Mun, S.-H. Yang, K. Ibrahim, Y. Tomioka, and C. S. Fadley, Phys. Rev. Lett. **92**, 166401 (2004).
- ²⁴ Y. Ishida, H. Ohta, A. Fujimori, and H. Hosono, J. Phys. Soc. Jpn. 76, 103709 (2007).
- ²⁵P. M. Chaikin and G. Beni, Phys. Rev. B **13**, 647 (1976).
- ²⁶ W. Koshibae, K. Tsutsui, and S. Maekawa, Phys. Rev. B **62**, 6869 (2000).